Course title: Engineering Thermodynamics

Course Basic Information						
Academic Unit:	Civil Engineering	g Faculty				
Course title:	Engineering Thermodynamics					
Level:	BSc					
Course Status:	Mandatory					
Year of Study:	II I					
Number of Classes per Week:	2+2					
ECTS Credits:	6 ECTS	6 ECTS				
Time /Location:	8:15-10:00,lab.	L 9				
Teacher:	Ali Muriqi	Ali Murigi				
Contact Details:	ali.muriqi@uni-	pr.edu, alimuriqi@	gmail.com, tel. 044			
	141 029					
Course Description:	The subject of Thermodynamics of Engineering studies the Law I of thermodynamics and the Law II of thermodynamics and deals with the way of determining the properties of homogeneous substances. Also the subject matter is the determination of the balance of mass, energy and entropy in solving various engineering problems.					
Course Goals:	The emphasis of the course will be on developing the ability of student for: 1. Application of the principle of mass and energy conservation for the evaluation of the performance of simple engineering systems 2. Analysis of thermodynamic properties of simple homogeneous substances 3. Analysis of processes and cycles using the Law II of thermodynamics to determine the maximum efficiency 4. Evaluation of wet air properties 5. Analysis of air conditioning processes 6. Analyzing the regularities of heat transfer to solve simple energy conversion problems					
Expected Learning Outcomes:	be able to: 1. Kr from the course	After completing this course (subject) the student will be able to: 1. Know the subject 2. Receive assistance from the course for drafting elaborations (conceptual projects) and professional projects				
Student Workload (should be						
Activity		Day/ Week	Total			
Lectures	2	15	30			
Theory/ Lab Work/Exercises	2	15	30			
Practical Work						
Consultations with the teaher						

Field Work		2	15	20	
Test, seminar paper		-			
Homework				5	
Self-study (library or home)		1	15	15	
Preparation for final exam				15	
Assessment time (test, quiz, final				30	
exam)					
Projects, presentations, etc		1	5	5	
Total		-		150	
Teaching Methods:		Lectures, exe	rcises during o	class using different	
		materials, one project work in group of 2-3 students			
		(independent work), individual homework			
Assessment Methods:		Individual assignments completed in class 30%;			
		Individual assignments completed at home 30%;			
		Exam 40%.			
Primary Literature:		Termodinamika e inxhinierisemligjëratat e			
		përgatitur nga Prof. Dr.Ali Muriqi,2020			
		2. Permbledhje detyrash nga Termodinamika,			
		Krasniqi, F. dhe			
		Muriqi, A., UP, Prishtinë 1996			
Additional Literature:		1. Demneri I .etj. Termodinamika , UPT, Tirane,2005			
			Çengel, Y.; Boles, D.: Thermodinamics- an Engineering Appraoch, McGraw Hill, 2011		
Designed to achine plan		Engineerin	g Appraoch, McGra	W HIII, 2011	
Designed teaching plan	Title of	the Leetuve			
Week		the Lecture			
Week 1:	Introduction to the basic concept of the course				
Week 2:	Basic state parameters				
Week 3:	Equation of state of gases				
Week 4:	Outdoor	Outdoor work, Heat, Specific Thermocapacity			
Week 5:	Law I of thermodynamics				
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Designed teaching	Pidii
Week	Title of the Lecture
Week 1:	Introduction to the basic concept of the course
Week 2:	Basic state parameters
Week 3:	Equation of state of gases
Week 4:	Outdoor work, Heat, Specific Thermocapacity
Week 5:	Law I of thermodynamics
Week 6:	Internal energy, Entalpia
Week 7:	Law II of thermodynamics. Reversible and irreversible processes
Week 8:	Carno Cycle
Week 9:	The polytropic process of state change.
Week 10:	Chemical thermodynamics and combustion processes.
	Stehiometric equations
Week 11:	The required amount of combustion air and the amount of
	combustion gases
Week 12:	Moliere's diagram and special cases
Week 13:	Humid air
Week 14:	Heat and mass transmission
Week 15:	Heat losses

Academic Policies and Code of Conduct

We start and finish class on time.

Tools used during class must be cleaned and stored away at the end of class.

Mobile/smart phones, and other electronic devices (e.g. iPods) must be turned off (or on vibrate) and hidden from view during class time.

Laptop and tablet computers are allowed for quiet use only; other activities such as checking personal e-mail or browsing the Internet are prohibited.